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Note: Not all of the reactions will occur. For those that do not, write no reaction. 1. A piece of copper is dropped into a container of water. No reaction 2. Liquid bromine is added to a container of sodium iodide crystals.  $\text{Br}_2(l) + 2\text{NaI}(s) \rightarrow 2\text{NaBr}(s) + \text{I}_2(s)$  3. An aluminum strip is immersed in a solution of silver nitrate.

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Re-typed from The Ultimate Chemical Equations Handbook by Hague and Smith 7. Free elements, no matter how complex the molecule, have an oxidation number (valence or charge) equal to zero. OJ. Fe. T~ He < /1 1 ~~ L.td. 8. The following are diatomic or polyatomic elements in nature which must be committed to memory.

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ammonium oxalate 4 224 polonium(VI) thiocyanate  $\text{Po}(\text{SCN})_6$  tetraphosphorus decaoxide  $\text{P}_4\text{O}_{10}$   
12. 13. 15. 16. 17. 18. 19. manganese(VII) oxide copper(II) dihydrogen phosphate

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