

## Cracking The Periodic Table Code Pogil Answer

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### Cracking The Periodic Table Code

Cracking the Periodic Table Code 3 7. Obtain the Electron Energy Levels handout from your teacher. Put the Electron Energy Levels handout next to Model 1. Start at the bottom of the Electron Energy Levels handout with the 1s energy sublevel and locate the section of the periodic table corresponding to that sublevel. Why

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POGIL rewrite for Cracking the Periodic Table Code Cracking the Periodic Table Code Why aren't the elements listed in alphabetical order? Why? ----- As charts go, the periodic table is a bit odd. It's not square. Large portions of the table appear to be missing at the top. It's not organized alphabetically so elements can be found easily.

### **Cracking the Periodic Table Code**

Date: \_\_\_\_ Period: \_\_\_\_ POGIL: Cracking the Periodic Code Why? As charts go, the periodic table is a bit odd. It's not square and large portions of the table appear to be 'missing' at the top. It's not organized alphabetically, so it can be hard to find elements. But to a chemist, the periodic table is a very powerful tool.

### **Name: Chemistry POGIL: Cracking the Periodic Code**

CrackingthePeriodicTableCode3. Model 2 - PeriodicTable(Long Form)123456711. Use your group's reconstructed periodic table to label thesections of Model 2.12. Compare theperiodic table of Model 1 with theperiodic table of Model 2.

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POGIL Activity: Cracking the Periodic Table Code Summary Note to Students: The following paragraphs are meant to summarize the activity you did in class. Compare this summary to the one your teacher asked you to write. For homework use your comparison to write down all changes you would need to make to your summary to make it accurate and complete.

### **POGIL Activity: Cracking the Periodic Table Code**

CRACKING THE PERIODIC TABLE CODE Author: SVSD Created Date: 10/17/2016 12:43:03 PM ...

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Unit 2 Module B: Chemical Symbols / Introduction to Periodic Table Chapter/Section Topic Ch. 6: 6.1, 6.2, 6.3 6.1 Organizing the elements 6.2 Classifying the elements 6.3 Periodic Trends Next Generation Science Standards HS-PSI-1 Use the periodic table as a model to predict the relative properties of elements based on the

### **CHEMISTRY QLM Q2**

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### **Cracking The Periodic Table Code Pogil Answer Key ...**

A A B d<sup>5</sup> c p<sup>4</sup> D 1W) Model 3 — Use of the Periodic Table for Electron Energy Levels Cracking the Periodic Table Code 5 18. Now that you understand how the structure of the periodic table relates to electron sublevels, you can use it as a “cheat sheet” for the order of filling of electrons in the sublevels of the atom.

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